

Synthesis of 3-Cyclohexanecarbonyl-2(3*H*)-Benzoxazolone and 3-Cyclohexanecarbonyl-2(3*H*)-Benzothiazolone Derivatives

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Abstrat.

In this study, 3- cyclohexanecarbonyl-2(3*H*)-benzoxazolone and 3- cyclohexanecarbonyl-2(3*H*)-benzothiazolone derivatives have been synthesized in excellent yields by condensation of 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone derivatives with cyclohexanecarbonyl acid chloride using sodium ethylate in a mixture of acetone-water (9/1) as solvent. The structures of synthesized derivatives were identified by FT-IR, ¹H-NMR and ¹³C-NMR spectroscopy.

Keywords: 2(3*H*)-benzoxazolone, 2(3*H*)-benzothiazolone, FC-acylation, Cyclohexanecarbonyl acid chloride, *N*-acylation.

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Introduction

2(3*H*)-benzoxazolone and its corresponding sulfur bioisoster 2(3*H*)-benzothiazolone constitute an essential class of heterocyclic compounds, due to their various Biological properties [1-13]. Many derivatives bearing various substituents have been reported to have different biological activities [14-22].

In particular, 6-Benzoyl-2(3*H*)-benzoxazolone (10194 CERM) and 6-benzoyl-2(3*H*)-benzothiazolone (S-14080) with analgesic and anti-inflammatory activities, were tested clinically as potential surrogates of the classical non-steroidal anti-inflammatory agent. They were shown to exhibit a peculiar pharmacological profile that acts not only as an inhibitor of COX-1 and COX-2 but also promotes the release of an opioid peptide (endomorphin) in the periphery [5,6].

In this connection and in an effort to generate dual mode of action analgesics endowed with COX inhibition, we designed a new series of 3-cyclohexanecarbonyl-2(3*H*)-benzoxazolones and 3-cyclohexanecarbonyl-2(3*H*)-benzothiazolones of general structure Figure 1.

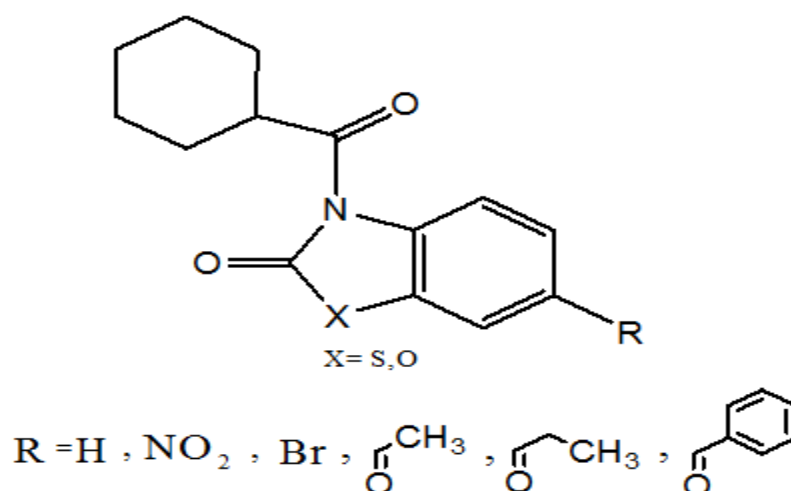


Figure 1 : General structure of the new series

Experimental

General

All reactions were monitored by thin layer chromatography using Merk TLC silica gel 60 F 254 and visualized with UV light, using Ethyl acetate: cyclohexane (7:3 v/v), a solvent system as eluent. Melting points were measured in open capillary tubes on an Electro thermal apparatus. The IR spectra were obtained on a Perkin-Elmer 457 spectrometer using KBr pellets. 1H and ^{13}C NMR spectra were recorded on a Bruker 400 spectrometer using DMSO- D_6 or $CDCl_3$ as solvents and (TMS) as an internal standard.

Methods and procedures

Synthesis of 6-nitro-2(3*H*)-benzoxazolone and 6-nitro-2(3*H*)-benzothiazolone (1b-2b)

To nitric acid (68%, 10 cm³, 150 mmol) cooled to -0-5°C was added dropwise, (10 mmol) of 2(3*H*)-benzoxazolone or 2(3*H*)-benzothiazolone. The reaction mixture was heated at 50°C for 30 min and kept at room temperature for 2h. After that, 100 cm³ of ice water was added to the reaction mixture and stirred for 15 min. The precipitate was filtered, washed with water, dried and recrystallized from ethanol to afford the corresponding 6-nitro-2(3*H*)-benzoxazolone and 6-nitro-2(3*H*)-benzothiazolone.

Synthesis of 6-bromo-2(3*H*)-benzoxazolone and 6-bromo-2(3*H*)-benzothiazolone 1b-2b

(0.004 mol) of 2(3*H*)-benzoxazolone or 2(3*H*)-benzothiazolone was dissolved under stirring in 50 ml of dichloromethane/tetrahydrofuran (8/2). After that, (6.87g, 0.043 mol) of Br_2 dissolved in a small amount of dichloromethane was slowly added. The reaction mixture was heated at 60°C for 2 hours and kept at room temperature for 48 hours. After evaporation of the solvent, the residue triturated with water, filtered, dried, and crystallized from ethanol to afford the corresponding 6-bromo-2(3*H*)-benzoxazolone and 6-bromo-2(3*H*)-benzothiazolone. The physical properties (m.p, IR, 1H NMR) are in accordance with published data [6, 25].

Synthesis of 6-acyl-2(3*H*)-benzoxazolones and 6-acyl-2(3*H*)- benzothiazolone derivatives, using FeCl₃-DMF complex (1c-6c).

(4,7 ml, 0.066 mol) of Dimethylformamide was added to ferric chloride (38.92 g, 0.24 mol). The mixture was then placed in an oil bath and heated at 90°C for 15 min. 2(3*H*)-benzoxazolone or 2(3*H*)-benzothiazolone (60 mmol) was added in portions. When the mixture was uniformly mixed, acid (6,4 ml, 0.06 mol) was added. Then the reaction mixture was heated at 170°C under stirring for 3 h. The products were poured into ice water with HCl (5ml, 37%). The precipitated was collected by filtration, dried and crystallized.

General procedure for the synthesis of 3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone and 3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone derivatives (1d-12d).

(0.004 mol) of 2(3*H*)-benzoxazolone or 2(3*H*)-benzothiazolone or their derivatives was dissolved under stirring in 50 ml of absolute ethanol. After that, (2.72g, 0.040 mol) of sodium ethylate was added under stirring to insure the formation of sodium derivatives. After evaporation of ethanol, 50 ml of acetone and 5 ml of water were added and then (5.44 ml, 0.041 mol) of cyclohexanecarbonyl acid chloride was slowly added. The mixture was stirred at room temperature for 15 min. After evaporation of the acetone, the residue triturated with a solution of 10% of (NaHCO₃), filtered, washed with water and dried, and crystallized from ethanol to afford good purity of the compounds with a yield (92-99%).

3-(Cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (1d). Yields: 99%. M.p: 84-85°C. IR (KBr, cm⁻¹): 2930 (νCH₂), 1805 (νC=O), 1723 (νC=O), 1592.09 (νC=Caromatic). ¹H-NMR (400 MHz, CDCl₃): δ=1.24-1.59 (m, 6H), 1.68-1.94 (m, 4H), 3.69 (m, 1H), 7.11-7.38 (m, 3H, Ar-H), 8.07 (d, *J* = 8.0 Hz, 1H, Ar-H). ¹³C-NMR (400 MHz, CDCl₃): δ= 175.82 (C=O), 151.33 (C=O), 142.33, 128.11, 125.15, 124.75, 116.43, 109.78 (aromatic carbons), 43.46 (CH), 28.92 (CH₂), 25.77 (CH₂), 25.45 (CH₂).

6-Bromo-3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (2d). Yields: 98%. M.p: 125-126°C. IR (KBr, cm⁻¹): 2932 (νCH₂), 1789 (νC=O), 1718 (νC=O), 1600 (νC=Caromatic). ¹H-NMR (400 MHz, CDCl₃): δ=1.19-1.55 (m, 6H), 1.54-1.96 (m, 4H), 3.58 (m, 1H), 7.16-7.31 (m, 2H, Ar-H), 7.89 (d, *J* = 7.9 Hz, 1H, Ar-H). ¹³C-NMR (400 MHz, CDCl₃): δ= 175.58 (C=O), 150.41 (C=O), 142.71, 130.84, 129.35, 127.80, 117.33, 113.48 (aromatic carbons), 43.47 (CH), 28.86 (CH₂), 25.64 (CH₂), 25.26 (CH₂).

6-Nitro-3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (3d). Yields: 94%. M.p: 123-124°C. IR (KBr, cm⁻¹): 2928 (νCH₂), 1790 (νC=O), 1718 (νC=O), 1600 (νC=Caromatic), 1338.49 (νNO₂). ¹H-NMR (400 MHz, DMSO-d₆): δ=1.25-1.58 (m, 6H), 1.67-1.95 (m, 4H), 3.71 (m, 1H), 8.08 (d, *J* = 7.91 Hz, 1H, Ar-H), 8.20-8.27 (m, 2H, Ar-H). ¹³C-NMR (400 MHz, DMSO-d₆): δ= 175.93 (C=O), 152.90 (C=O), 142.29, 128.56, 126.15, 125.15, 117.32, 110.28 (aromatic carbons), 43.56 (CH), 28.96 (CH₂), 25.87 (CH₂), 25.55 (CH₂).

6-Acetyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (4d). Yields: 95%. M.p: 107-108°C. IR (KBr, cm⁻¹): 2928 (νCH₃), 1760 (νC=O), 1719 (νC=O), 1676 (νC=O), 1600

($\nu_{\text{C=Caromatic}}$). $^1\text{H-NMR}$ (400 MHz, DMSO-d_6): δ =1.24-1.56 (m, 6H), 1.64-1.93 (m, 4H), 2.68 (s, 3H), 3.86 (m, 1H), 7.75-7.86 (m, 2H, Ar-H), 8.17 (d, J = 7.76 Hz, 1H, Ar-H). $^{13}\text{C-NMR}$ (400 MHz, DMSO-d_6): δ = 198.62 (C=O), 174.72 (C=O), 152.79 (C=O), 142.43, 132.34, 129.75, 123.63, 109.63, 108.85 (aromatic carbons), 43.36 (CH), 28.94 (CH_3), 25.74 (CH_2), 25.52 (CH_2), 24.46 (CH_2).

6-Propionyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (5d). Yields: 93%. M.p: 149-150°C. IR (KBr, cm^{-1}): 2931 (ν_{CH_3}), 1780 ($\nu_{\text{C=O}}$), 1724 ($\nu_{\text{C=O}}$), 1683 ($\nu_{\text{C=O}}$), 1609 ($\nu_{\text{C=Caromatic}}$). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ =1.22-1.55 (m, 9H), 1.65-1.97 (m, 4H), 2.92 (q, 2H), 3.84 (m, 1H), 7.76-7.85 (m, 2H, Ar-H), 8.17 (d, J = 7.79 Hz, 1H, Ar-H). $^{13}\text{C-NMR}$ (400 MHz, CDCl_3): δ = 198.64 (C=O), 175.82 (C=O), 151.78 (C=O), 142.23, 133.30, 130.75, 124.63, 115.86, 109.25 (aromatic carbons), 43.35 (CH), 31.82 (CH_2), 28.85 (CH_2), 25.76 (CH_2), 25.54 (CH_2), 8.59 (CH_3).

6-Benzoyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolone (6d). Yields: 92%. M.p: 135-136°C. IR (KBr, cm^{-1}): 2933 (ν_{CH_2}), 1790 ($\nu_{\text{C=O}}$), 1725 ($\nu_{\text{C=O}}$), 1652 ($\nu_{\text{C=O}}$), 1600.09 ($\nu_{\text{C=Caromatic}}$). $^1\text{H-NMR}$ (400 MHz, DMSO-d_6): δ =1.26-1.57 (m, 6H), 1.65-1.97 (m, 4H), 3.84 (m, 1H), 7.40-7.73 (m, 7H), 8.14 (d, J = 7.71 Hz, 1H). $^{13}\text{C-NMR}$ (400 MHz, DMSO-d_6): δ = 193.58 (C=O), 176.78 (C=O), 151.18 (C=O), 141.73, 137.06, 134.50, 133.47, 130.97, 129.79, 128.36, 127.64, 115.42, 111.16 (aromatic carbons), 44.06 (CH), 27.92 (CH_2), 25.75 (CH_2), 25.35 (CH_2).

3-(Cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (7d). Yields: 99%. M.p: 48-49°C. IR (KBr, cm^{-1}): 2938 (ν_{CH_2}), 1710 ($\nu_{\text{C=O}}$), 1665 ($\nu_{\text{C=O}}$), 1598 ($\nu_{\text{C=Caromatic}}$). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ =1.17-1.53 (m, 6H), 1.65-1.96 (m, 4H), 3.56 (m, 1H), 7.16-7.39 (m, 3H, Ar-H), 8.04 (d, J = 7.96 Hz, 1H, Ar-H). $^{13}\text{C-NMR}$ (400 MHz, CDCl_3): δ = 177.36 (C=O), 170.22 (C=O); 134.85; 126.24; 124.87; 122.96; 121.81; 111.31 (aromatic carbons). 44.13 (CH); 28.62 (CH_2); 25.54 (CH_2); 25.31 (CH_2).

6-Bromo-3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (8d). Yields: 98%. M.p: 81-82°C. IR (KBr, cm^{-1}): 2935 (ν_{CH_2}), 1725 ($\nu_{\text{C=O}}$), 1677 ($\nu_{\text{C=O}}$), 1595 ($\nu_{\text{C=Caromatic}}$). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ =1.18-1.56 (m, 6H), 1.53-1.98 (m, 4H), 3.57 (m, 1H), 7.16-7.28 (m, 2H, Ar-H), 7.91 (d, J = 7.89 Hz, 1H, Ar-H). $^{13}\text{C-NMR}$ (400 MHz, CDCl_3): δ = 175.24 (C=O), 169.18 (C=O), 134.75, 131.37, 127.49, 125.54, 117.57, 110.89 (aromatic carbons), 44.86 (CH), 29.10 (CH_2), 25.52 (CH_2), 25.14 (CH_2).

6-Nitro-3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (9d). Yields: 92%. M.p: 137-138°C. IR (KBr, cm^{-1}): 2934 (ν_{CH_2}), 1715 ($\nu_{\text{C=O}}$), 1672 ($\nu_{\text{C=O}}$), 1600 ($\nu_{\text{C=Caromatic}}$), 1338.49 (ν_{NO_2}). $^1\text{H-NMR}$ (400 MHz, CDCl_3): δ =1.16-1.56 (m, 6H), 1.65-1.98 (m, 4H), 3.58 (m, 1H), 8.15 (d, J = 7.79 Hz, 1H, Ar-H), 8.19-8.25 (m, 2H, Ar-H). $^{13}\text{C-NMR}$ (400 MHz, CDCl_3): δ = 174.34 (C=O), 169.62 (C=O), 137.36, 131.56, 124.34, 122.86, 121.71, 110.81 (aromatic carbons), 43.70 (CH), 27.08 (CH_2), 25.84 (CH_2), 24.88 (CH_2).

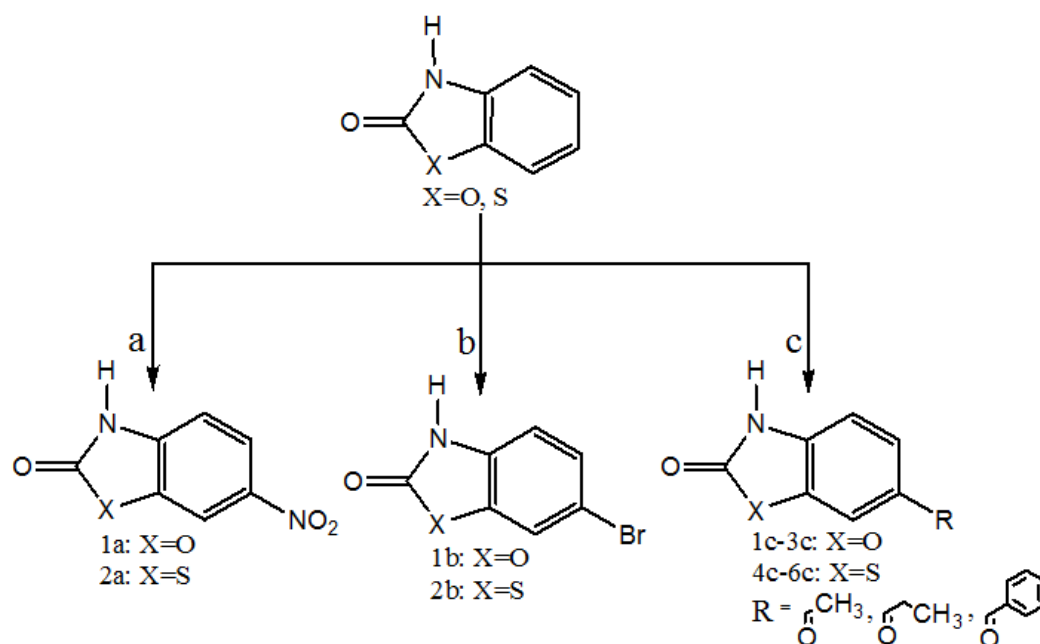
6-Acetyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (10d). Yields: 95%. M.p: 99-100°C. IR (KBr, cm⁻¹): 2937 (νCH₃), 1710 (νC=O), 1683 (νC=O), 1675 (νC=O), 1598 (νC=Caromatic). ¹H-NMR (400 MHz, DMSO-d₆): δ=1.24-1.59 (m, 6H), 1.63-1.96 (m, 4H), 2.64 (s, 3H), 3.85 (m, 1H), 7.93-8.10 (m, 2H, Ar-H), 8.14 (d, *J* = 7.86 Hz, 1H, Ar-H). ¹³C-NMR (400 MHz, DMSO-d₆): δ= 197.62 (C=O), 176.72 (C=O), 169.79 (C=O), 140.43, 135.34, 128.75, 123.73, 122.93, 117.85 (aromatic carbons), 43.46 (CH), 28.97 (CH₃), 25.69 (CH₂), 25.49 (CH₂), 24.47 (CH₂).

6-Propionyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (11d). Yields: 96%. M.p: 103-104°C. IR (KBr, cm⁻¹): 2935 (νCH₃), 1728 (νC=O), 1680 (νC=O), 1677 (νC=O), 1610 (νC=Caromatic). ¹H-NMR (400 MHz, DMSO-d₆): δ= 1.22-1.56 (m, 9H), 1.63-1.98 (m, 4H), 2.90 (q, 2H), 3.89 (m, 1H), 7.86-7.98 (m, 2H, Ar-H), 8.15 (d, *J* = 7.89 Hz, 1H, Ar-H). ¹³C-NMR (400 MHz, DMSO-d₆): δ= 194.64 (C=O), 177.82 (C=O), 169.98 (C=O), 139.23, 133.25, 127.15, 123.63, 121.86, 116.25 (aromatic carbons), 43.33 (CH), 32.26 (CH₂), 28.82 (CH₂), 25.69 (CH₂), 25.54 (CH₂), 8.29 (CH₃).

6-Benzoyl-3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolone (12d). Yields: 93%. M.p: 127-128°C. IR (KBr, cm⁻¹): 2934 (νCH₂), 1695 (νC=O), 1670 (νC=O), 1648 (νC=O), 1595.10 (νC=Caromatic). ¹H-NMR (400 MHz, DMSO-d₆): δ=1.25-1.58 (m, 6H), 1.64-1.96 (m, 4H), 3.87 (m, 1H), 7.43-7.82 (m, 7H), 8.13 (d, *J* = 8.01 Hz, 1H). ¹³C-NMR (400 MHz, DMSO-d₆): δ= 196.58 (C=O), 174.78 (C=O), 165.88 (C=O), 140.73, 137.25, 134.30, 132.79, 129.97, 129.39, 128.46, 127.60, 123.42, 116.66 (aromatic carbons), 44.16 (CH), 28.12 (CH₂), 25.82 (CH₂), 25.36 (CH₂).

Results and discussions

The 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone derivatives used as intermediates were synthesized via the route outlined in Scheme 1. The classical nitration of the aromatic ring of 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone with nitric acid produced the 6-nitro-2(3*H*)-benzoxazolone and 6-nitro-2(3*H*)-benzothiazolone **1a-2a**, in 68 to 70% yields [29, 30]. The bromination procedure using Br₂ in CH₂Cl₂/THF (8/2) yielded the corresponding 6-bromo-2(3*H*)-benzoxazolone and 6-bromo-2(3*H*)-benzothiazolone **1b-2b** in 86 and 89% yields after 48h [31].

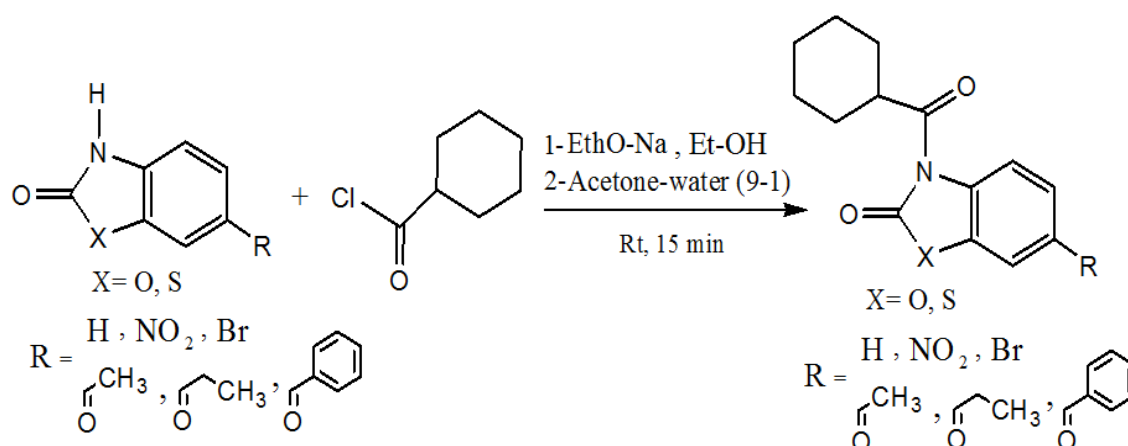


Scheme 1: Reagents and conditions: (a) HNO_3 (68%), $-5-0^\circ\text{C}$; (b) Br_2 , $\text{CH}_2\text{Cl}_2/\text{THF}$, 60°C 1h, 48h

rt; (c) $\text{FeCl}_3\text{-DMF}$, RCOCl , 170°C .

The 6-acyl-2(3*H*)-benzoxazolone and 6-acyl-2(3*H*)-benzothiazolone derivatives (1c-6c) were obtained in satisfactory yields (67-75%) using acid chlorides as acylating agents in the presence of $\text{FeCl}_3\text{-DMF}$ complex used as Friedel-Crafts catalyst [12].

The 3-cyclohexanecarbonyl-2(3*H*)-benzoxazolone and 3-cyclohexanecarbonyl-2(3*H*)-benzothiazolone derivatives (1d-12d) were obtained after 15 min at room temperature by a series of reactions from 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone derivatives with cyclohexanecarbonyl acid chloride, using sodium ethylate instead of sodium hydroxide (Scheme 2) [6].



Scheme 2: Synthesis of 3-cyclohexanecarbonyl-2(3*H*)-benzoxazolone and 3-cyclohexanecarbonyl-

2(3*H*)-benzothiazolone derivatives (1d-12d)

The synthesized products were obtained with a very high purity, which was characterized by thin-layer chromatography. The yields of the synthesized compounds are presented in Table 1.

Table 1. Physicochemical data of the synthesized 3-(cyclohexanecarbonyl)-2(3*H*)-Benzoxazolone

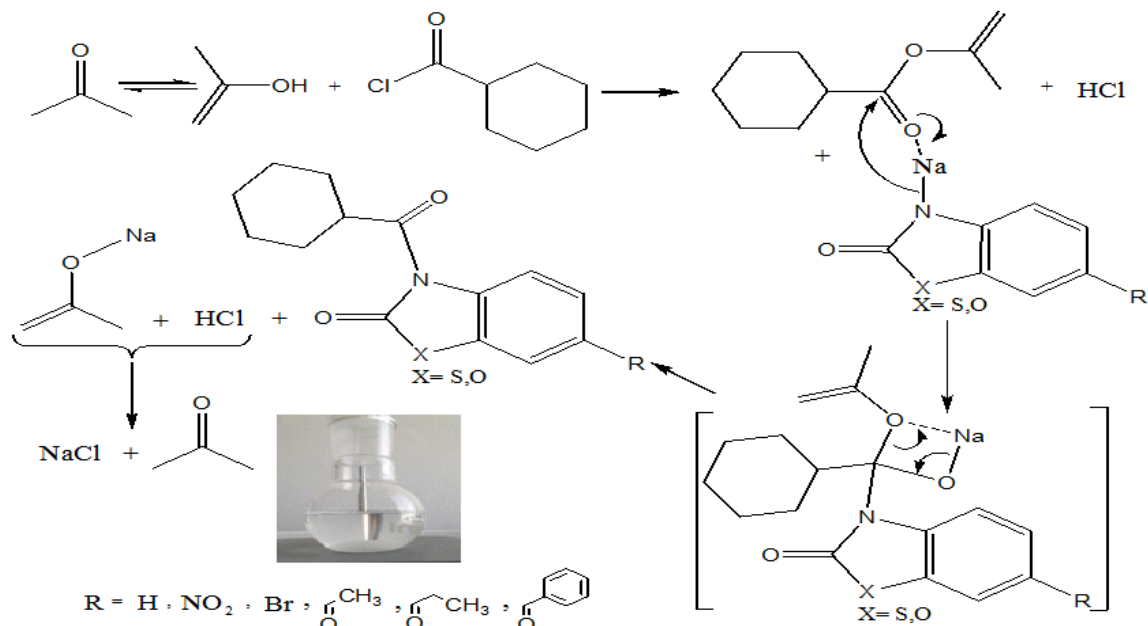
and 3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolones compounds 1d-12d.

Product	X	R	^a Mp (C)	Time (min)	^b Yield (%)	^c Mol. F
1d	O	H	84-85	15	99	C ₁₄ H ₁₅ NO ₃
2d	O	Br	125- 126	15	98	C ₁₄ H ₁₄ NO ₃ Br
3d	O	NO ₂	123- 124	15	94	C ₁₄ H ₁₄ N ₂ O ₅
4d	O	C ₆ H ₅	135- 136	15	92	C ₂₁ H ₁₉ NO ₄
5d	O	CH ₃	107- 108	15	95	C ₁₆ H ₁₇ NO ₄
6d	O	CH ₂ CH ₃	149- 150	15	93	C ₁₇ H ₁₉ NO ₄
7d	S	H	48-49	15	99	C ₁₄ H ₁₅ NO ₂ S
8d	S	Br	81-82	15	98	C ₁₄ H ₁₄ NO ₂ SBr
9d	S	NO ₂	137- 138	15	92	C ₁₄ H ₁₄ N ₂ O ₄ S
10d	S	C ₆ H ₅	127- 128	15	93	C ₂₁ H ₁₉ NO ₃ S
11d	S	CH ₃	99-100	15	95	C ₁₆ H ₁₇ NO ₃ S
12d	S	CH ₂ CH ₃	103- 104	15	96	C ₁₇ H ₁₉ NO ₃ S

^aMelting point; ^bIsolated yield after crystallization; ^cMolecular formula

Under the experimental conditions, we observed a transformation of the heterogeneous reaction medium to a homogeneous phase, after the addition of cyclohexanecarbonyl chloride, and after

that the precipitation of sodium chloride (NaCl) as a white powder at the end of the reaction. This transformation can be expounded by the formation of an intermediate according to the proposed reaction mechanism reported in Scheme 3.



Scheme 3: the proposed mechanism of the formation of 3-cyclohexanecarbonyl derivatives

The addition of cyclohexanecarbonyl acid chloride leads to the formation of the isopropenyl cyclohexane carboxylate by reaction with the enol form of acetone, whose reacted with the sodium derivatives of 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone, leading to the formation of the intermediate, and after rearrangement, conducted to the formation of *N*-acyl compounds and the sodium chloride NaCl at the end of the reaction.

Conclusions

Through this work, we confirmed the efficacy of the experimental protocol using the sodium ethylate in the mixture of acetone water for the synthesis of *N*-acyl derivatives. These mild conditions leads to the formation of the isopropenyl cyclohexane carboxylate intermediate, and after rearrangement permitted rapid access to 3-(cyclohexanecarbonyl)-2(3*H*)-benzoxazolones and 3-(cyclohexanecarbonyl)-2(3*H*)-benzothiazolones derivatives with good yields.

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